Structure-Based Function Discovery of an Enzyme for the Hydrolysis of Phosphorylated Sugar Lactones

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Supporting Information

ABSTRACT: Two enzymes of unknown function from the cog1735 subset of the amidohydrolase superfamily (AHS), LMO2365_2620 (Lmo2620) from Listeria monocytogenes str. 4b F2365 and Bh0225 from Bacillus halodurans C-125, were cloned, expressed, and purified to homogeneity. The catalytic functions of these two enzymes were interrogated by an integrated strategy encompassing bioinformatics, computational docking to three-dimensional crystal structures, and library screening. The three-dimensional structure of Lmo2620 was determined at a resolution of 1.6 Å with two phosphates and a binuclear zinc center in the active site. The proximal phosphate bridges the binuclear metal center and is 7.1 Å from the distal phosphate. The distal phosphate hydrogen bonds with Lys-242, Lys-244, Arg-275, and Tyr-278. Enzymes within cog1735 of the AHS have previously been shown to catalyze the hydrolysis of substituted lactones. Computational docking of the high-energy intermediate form of the KEGG database to the three-dimensional structure of Lmo2620 highly enriched anionic lactones versus other candidate substrates. The active site structure and the computational docking results suggested that probable substrates would likely include phosphorylated sugar lactones. A small library of diacid sugar lactones and phosphorylated sugar lactones was synthesized and tested for substrate activity with Lmo2620 and Bh0225. Two substrates were identified for these enzymes, D-lyxono-1,4-lactone-5-phosphate and L-ribono-1,4-lactone-5-phosphate. The $k_{cat}/K_m$ values for the cobalt-substituted enzymes with these substrates are $\sim$10$^3$ M$^{-1}$ s$^{-1}$.

The determination of function of uncharacterized enzymes remains challenging for the millions of new proteins identified by large scale DNA sequencing efforts because there are currently no reliable high-throughput methods for deciphering enzyme specificity. Thus, the mapping of new reactions and metabolic pathways has not kept pace with the identification of new genes within completely sequenced bacterial genomes. An assessment of the functional annotations for the more than 12 million genes thus far sequenced suggests that approximately one-third of the encoded proteins have an uncertain, unknown, or incorrect function.1−5 This implies that a substantial portion of the metabolic landscape remains to be identified. We have therefore explored developing comprehensive strategies for annotating enzymes of unknown function through a combination of bioinformatics, computational docking of potential ligands to protein structures, and synthesis of focused compound libraries for experimental verification. We have applied these methods to the elucidation of function for enzymes within the amidohydrolase superfamily.6−20

The amidohydrolase superfamily (AHS) was first recognized by Sander and Holm on the basis of the structural similarities of urease, phosphotriesterase, and adenosine deaminase.21 Enzymes within this superfamily catalyze hydrolytic reactions of amide and ester bonds contained within carbohydrates, peptides, and nucleic acids.22 Less common amidohydrolase-catalyzed reactions include decarboxylations, hydrations, and isomerizations.11,23−25 All proteins within the AHS have a distorted (β/α)$_9$-barrel fold and contain either a mononuclear or a binuclear metal center in the active site. The metal center activates a hydrolytic water molecule for nucleophilic attack and enhances the electrophilic character of the target substrate. More than 12000 unique protein sequences have been identified as members of the AHS within the genomes of the first 1000 completely sequenced bacteria deposited in the National Center for Biotechnology Information (NCBI). Proteins identified in the AHS have been subdivided into 24 clusters of orthologous groups (COG).26

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Enzymes in cog1735 have been reported to catalyze the hydrolysis of organophosphates and lactones. Shown in Figure 1 is a sequence similarity network analysis representation of this COG at a BLAST E value cutoff of $1 \times 10^{-30}$. The bacterial phosphotriesterase (PTE) from group 8 catalyzes the hydrolysis of a variety of organophosphates, including the insecticide paraoxon and the chemical warfare agents GB, GD, and VX. Three other proteins from group 9, Sso2522, from the hyperthermophilic Sulfolobus solfataricus, has a weak phosphotriesterase activity but hydrolyzes lactones at substantially faster rates. Three other proteins from group 3, Mt0240 from Mycobacterium tuberculosis, Rer5000 from Rhodococcus erythropolis, and MAP3668c from Mycobacterium avium, have been shown to catalyze the hydrolysis of N-acetyl homoserine lactones. A thermostable lactonase from Geobacillus stearothermophilus (gi3258588268) from group 7 has phosphotriesterase and lactonase activities. The PTE homology protein (PHP) from Escherichia coli K12 (gi3379) from group 1 was structurally characterized in 1998, but its catalytic function remains unknown. We, and others, have determined the three-dimensional structure and substrate profile of Dr0930 from Deinococcus radiodurans [Protein Data Bank (PDB) entries 3FDK and 3HTW] from group 7. This protein has very low phosphotriesterase activity but efficiently hydrolyzes β- and γ-lactones with an alkyl substitution at the carbon adjacent to the ring oxygen.

Here we combine structure-based docking screens of a general metabolite library with biochemical screens of a focused chemical library to determine the substrate profile for enzymes from group 5 of cog1735. LMOF2365_2620 (Lmo2620) from Listeria monocytogenes str. 4b F2365 and Bh0225 from Bacillus halodurans C-12S were purified to homogeneity, and the three-dimensional structure of Lmo2620 was determined at a resolution of 1.6 Å. These two proteins have 74% identical sequences and were found to be lactonases that catalyze the hydrolysis of d-lyxono-1,4-lactone-5-phosphate and L-ribo-1,4-lactone-5-phosphate.

### MATERIALS AND METHODS

**Materials.** The genomic DNA of L. monocytogenes str. 4b F2365 and that of B. halodurans C-125 were purchased from ATCC. Restriction enzymes and T4 DNA ligase were purchased from New England Biolabs. Expression vector pET30α(+) and E. coli strain BL21(DE3) were obtained from Novagen, and the Platinum Pfx DNA polymerase was obtained from Invitrogen. The Wizard Miniprep DNA purification kit was obtained from Promega. LB broth was purchased from Tpi Research Products International. Chromatographic gel filtration columns and Resource Q anion exchange columns were purchased from GE Healthcare. ICP standards were purchased from Inorganic Ventures. All other buffers, purification reagents, and chemicals used in this investigation were purchased from Sigma, unless otherwise stated.

**Synthesis of Sugar Lactone Phosphates.** The phosphorylated sugar lactones (compounds 1–10) were made by direct phosphorylation of the parent lactone or the 2,3-isopropylidene-protected lactone. d-Ribono-1,4-lactone was converted to 2,3-O-isopropylidene-d-ribo-1,4-lactone, and then 5.0 mmol of this material was reacted with diphenylchloroarsine (6.0 mmol) in pyridine, yielding S-(diphenylphosphoryl)-2,3-O-isopropylidene-d-ribo-1,4-lactone with a yield of 80%. This intermediate (4.0 mmol) was dissolved in 40 mL of acetic acid containing 50 mg of PtO2 and then H2 was applied overnight. After removal of the catalyst and solvent, a quantitative yield of 2,3-O-isopropylidene-d-ribo-1,4-lactone-5-phosphate was obtained. The protecting group was removed by the addition of a trifluoroacetic acid/dichloromethane mixture (9/1) for 4 h, yielding d-ribo-1,4-lactone-5-phosphate. Two equivalents of sodium bicarbonate was added, yielding the disodium salt: $^{31}P$ NMR (121.4 MHz, CDCl3) δ 3.36 (5%), 2.54 (95%), MS (ESI negative mode) found 226.99 (M−H)− and C$_6$H$_8$O$_3$P (M), computed 228.00. d-Lyxo-1,4-lactone-5-phosphate (1) was prepared in the same way from d-lyxono-1,4-lactone.

Compounds 2, 3, and 5–9 were synthesized similarly, but the isopropylidene protection step was not used. For the synthesis of compound 2, d-xylose (6.0 g, 40 mmol) and NaHCO$_3$ (5.04 g, 60 mmol) were stirred together in 50 mL of water at 0 ℃, and then Br$_2$ (3 × 0.76 mL, 44 mmol) was added at 15–20 min intervals. The mixture was stirred for 1 h and then for 3 h at room temperature. The solvent was removed, and the semisolid residue was extracted with ethanol (2 × 100 mL). After removal of the ethanol, the oily crude product (d-xylo-1,4-lactone) was obtained in a yield of 50%. The crude d-xylo-1,4-lactone (∼5 mmol) was mixed with diphenylchloroarsine (6.0 mmol) in 60 mL of pyridine and stirred overnight. The solvent was removed and the product purified by elution from a silica gel column with an ethyl acetate/acetone mixture (6/1), yielding S-diphenylphosphoryl-d-xylo-1,4-lactone in a yield of 50%. This compound (2.0 mmol) was hydrogenated in acetic acid with a PtO$_2$ catalyst overnight to produce d-xylo-1,4-lactone-5-phosphate. The disodium salt was made as described above: $^{31}P$ NMR (121.4 MHz, CDCl3) δ 2.96; MS (ESI negative mode) found 228.98 (M−H)− and C$_6$H$_8$O$_3$P (M), computed 228.00. Compounds 3 and 5–9 were obtained using the same procedure with similar results. All of the preparations had a 1–20% contamination of the hydrolyzed lactone. L-Ascorbate-6-phosphate...
(10) was prepared as described previously. The structures of these compounds are presented in Scheme 1.

**Synthesis of Dicarboxylate Sugar Lactones.** Compounds 11–18 were synthesized from xylitol, d-arabitol, l-arabitol, and adonitol. The alcohols were oxidized by nitric acid to the corresponding sugar acids and dehydrated to a mixture of 1,4- and 2,5-lactones. The major contaminant, oxalic acid, was removed at low temperatures and then the monoacid lactone converted to the sodium salt with NaHCO3. Oxidation of xylitol and conversion of the xylaric acid product to the monolactone provided a racemic mixture of compounds 11 and 15. A 2/1 mixture of compounds 12 and 16 was prepared by the oxidation of d-arabitol. Oxidation of l-arabitol gave a 2/1 mixture of compounds 13 and 17. The racemic mixture of compounds 14 and 18 was initiated by the oxidation of adonitol. All four preparations were contaminated with 10–15% of the diacid intermediate. The structures of compounds 11–18 are presented in Scheme 2.

The sugar alcohol (3 g of xylitol, d-arabitol, l-arabitol, or adonitol) was dissolved in concentrated nitric acid (12 mL), and then 80 mg of NaNO2 was added. The mixture was heated in an oil bath at 50 °C for 5–10 min. After the evolution of gas subsided, the solution was heated at 50–60 °C for an additional 16 h. The reaction mixture was allowed to cool to room temperature and then further cooled to −20 °C. The white crystalline precipitate (oxalic acid) was filtered, and the clear solution was cooled and filtered again. The solvent was removed by evaporation under reduced pressure at 70–75 °C for 5–6 h. This procedure allows for the removal of nitric acid and lactonization of acid sugars. The resulting products were dried for 24 h under vacuum at room temperature as colorless or light yellow solids. The lactone products were obtained in the yields of 50–60%.

**Xylaric acid 1,4- and 2,5-lactones (11 and 15, respectively):** 1H NMR (D2O) δ 4.41 (d, J = 8.1 Hz, 1H), 4.55 (t, J = 7.7 Hz, 1H), 5.01 (d, J = 7.5 Hz, 1H); 13C NMR (D2O) δ 70.96, 72.55, 77.21, 171.01, 175.91; TOF MS/ESI (M−H)− 161.47 (found), 162.02 (M, calcd).

**d-Arabinaric acid 1,4- and 2,5-lactones (12 and 16, respectively):** 1H NMR (D2O) δ 4.26 (t, J = 8.5 Hz, 0.5H), 4.43 (d, J = 8.7 Hz, 0.5H), 4.55–4.65 (m, 2.5H), 5.00 (d, J = 3.0 Hz); TOF MS/ESI (M−H)− 161.02 (found), 162.02 (M, calcd). d-Arabinaric acid 1,4-lactone (12): 13C NMR (D2O) δ...
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97.69, 67.90, 78.24, 169.65, 176.6. d-Arabinaric acid 2,5-lactone (16): 13C NMR (D2O) δ 73.04, 75.20, 76.79, 171.17, 174.5.

l-Arabinaric acid 1,4- and 2,5-lactones (13 and 17, respectively): 1H NMR (DMSO-d6) δ 4.07 (t, J = 7.5 Hz, 0.5H), 4.19 (d, J = 7.2 Hz, 0.5H), 4.41 (d, J = 3.3 Hz, 1H), 4.45 (d, J = 4.8 Hz, 1H), 4.93 (d, J = 3.0 Hz, 1H); TOF MS/ESI (M − H)− 161.02 (found), 162.02 (M, calcld).

Ribaric acid 1,4- and 2,5-lactones (14 and 18, respectively): 1H NMR (DMSO-d6) δ 4.25 (d, J = 5.1 Hz, 1H), 4.31 (d, J = 4.8 Hz, 1H), 4.74 (s, 1H); 13C NMR (DMSO-d6) δ 68.43, 71.24, 81.10, 170.00, 176.11; TOF MS/ESI (M − H)− 161.08 (found), 162.02 (M, calcld).

A racemic mixture of compounds 19 and 20 was synthesized from mucic acid.40 d-Gluaro-1,4-lactone (21) and d-gluaro-6,3-lactone (22) were synthesized from d-gluaric acid.41 Lactones were converted to their sodium salts with NaHCO3.

Cloning of Lmo2620 and Bh0225. The gene for Lmo2620 (gi|46908819) was amplified from the genomic DNA of L. monocytogenes str. 4b F2365. The amplified polymerase chain reaction (PCR) product was purified using a PCR cleanup system (Promega), digested with AseI and BamHI, and ligated to a pET30a(+) vector that had been Ndel and BamHI digested with Ndel and BamHI. The entire cloned fragment was sequenced to confirm the fidelity of the PCR amplification. Bh0225 (gi|15612788) was amplified from the genomic DNA of B. halodurans C-125. The amplified PCR product was purified using a PCR cleanup system (Promega), digested with Asel and BamHI, and ligated to a pET30a(+) vector that had been digested with Ndel and BamHI. The cloned fragment was sequenced to verify the fidelity of the PCR amplification.

Expression and Purification of Lmo2620 and Bh0225. To express Lmo2620, BL21(DE3) (Novagen) were transformed with the pET30a(+) plasmid containing the gene for Lmo2620. A single freshly transformed colony was cultured in LB medium supplemented with 50 µg/mL kanamycin at 37 °C. The overnight culture was inoculated into 1 L of LB medium and cultured at 30 °C with vigorous shaking. When the OD600 of the culture reached 0.6, the expression of the target protein was allowed to grow at room temperature for 18 h and then lysed by sonication (5 s pulses for 30 min) at 0 °C.

Centrifugation, the nucleic acids were removed via addition of 20 mL of a 2% (w/v) solution of protamine sulfate in 50 mM Hepes (pH 7.5). After centrifugation, the nucleic acids were removed via addition of 20 mL of a 2% (w/v) solution of protamine sulfate in 50 mM Hepes (pH 7.5). After centrifugation, the supernatant solution was fractionated via 40–60% saturation of ammonium sulfate. The precipitated protein was resuspended in 50 mM Hepes (pH 7.5), filtered with a 0.2 µm syringe filter (VWR), and loaded onto a HiLoad 26/60 Superdex-200 gel filtration column (Amersham Pharmacia). Protein was eluted at a flow rate of 2.0 mL/min. The fractions containing the target protein were pooled, loaded onto a Resource Q anion exchange column (Amersham Pharmacia), and then eluted with a linear gradient of NaCl in 20 mM Hepes (pH 7.5). The purity of the protein was estimated to be greater than 95% based on sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS–PAGE). The purified protein was designated as [Zn/Zn]-Lmo2620. Lmo2620 containing manganese or cobalt in the active site was obtained in good yield by adding 1.0 mM MnCl2 or CoCl2 during expression. The iron-specific chelator, 2,2′-bipyridyl, was added at a concentration of 0.1 mM to the growth medium when the absorbance reached 0.1 when the cells were grown in the presence of MnCl2 to reduce the iron content of the protein.22 These proteins are designated as [Mn/Mn]-Lmo2620 and [Co/Co]-Lmo2620, respectively.

Bh0225 was expressed and purified using the same protocol that was used for Lmo2620. Three forms of Bh0225 were prepared ([Zn/Zn]-Bh0225, [Mn/Mn]-Bh0225, and [Co/Co]-Bh0225) by supplementing the growth medium with 1.0 mM ZnCl2, MnCl2, or CoCl2 at the time of induction. Each form of Bh0225 was purified to homogeneity using ammonium sulfate fractionation (40–60% saturation), gel filtration chromatography, and anion exchange chromatography. The purity of Bh0225 was estimated to be greater than 95% by SDS–PAGE.

Metal Analysis. The protein concentration was determined spectrophotometrically at 280 nm using a SPECTRAMax-340 UV–vis spectrophotometer. The extinction coefficient at 280 nm used for calculating the concentration of Lmo2620 was 47455 M−1 cm−1, and the extinction coefficient at 280 nm for calculating the concentration of Bh0225 was 46490 M−1 cm−1 (ca.expasy.org/tools/protparam.html). The metal content of the as-purified proteins was determined by inductively coupled plasma emission mass spectrometry (ICP-MS). Before ICP-MS measurements, the protein samples were treated with concentrated nitric acid for 20 min at 100 °C and then diluted with distilled water until the final concentration of nitric acid was 1%.

Kinetic Measurements and Data Analysis. The kinetic measurements were conducted with a SPECTRAMax-340 plate reader. The hydrolysis of lactones was monitored using a pH-sensitive colorimetric assay.17 Protons released from the carboxylate product were measured using the pH indicator cresol purple. The reactions were performed in 2.5 mM Bicine (pH 8.3) containing 0.2 M NaCl, variable amounts of substrate, 0.1 mM cresol purple, and Lmo2620 or Bh0225 in 1% DMSO. The change in absorbance at 577 nm was monitored. The conversion factor (ε) was determined to be 1.76 × 105 M−1 cm−1 using acetic acid. The protein was stored in 50 mM Hepes (pH 7.5), and the buffer was exchanged with 10 mM Bicine (pH 8.3) using a PD-10 desalting column (GE Healthcare) before use. A background rate was observed in the absence of substrate because of the acidification by atmospheric CO2. The background rate was independent of substrate concentration and subtracted from the initial rates.

Kinetic parameters were obtained by fitting the initial rates directly to eq 1

\[
\nu / E_t = k_{cat}[A] / (K_m + [A])
\]

where \(\nu\) is the initial velocity, \(E_t\) is the enzyme concentration, \(k_{cat}\) is the turnover number, \([A]\) is the substrate concentration, and \(K_m\) is the Michaelis constant.

Crystallization and Data Collection. Crystals of Lmo2620 complexed with zinc and phosphate were grown by the hanging drop vapor diffusion method at room temperature by mixing equal volumes of protein and precipitant and equilibrating over the precipitant. The protein solution contained Lmo2620 (13.5 mg/mL) in 20 mM Hepes (pH 7.5), 0.1 M NaCl, and 1.0 mM ZnCl2; the precipitant contained 0.4 M Na2PO4, 1.6 M K3PO4, 0.1 M imidazole (pH 8.0), 0.2 M NaCl, and 1.0 mM ZnCl2. Crystals appeared in 3–4 days and exhibited diffraction consistent with space group P21, with six molecules of the protein complex per asymmetric unit (Table 1). Prior to data collection, crystals were transferred to cryoprotectant solutions composed of their mother liquids and 20% glycerol. After being
incubated for ∼10 s, the crystals were flash-cooled in a nitrogen stream. Diffraction data were collected at the NSLS X4A beamline (Brookhaven National Laboratory) on an ADSC CCD detector. Diffraction data were integrated and scaled with DENZO and SCALEPACK.\(^{43}\) The data collection statistics are given in Table 1.

**Table 1. Data Collection and Refinement Statistics for Lmo2620**

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<th>Data Collection</th>
<th>Refinement</th>
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<tr>
<td>beamline</td>
<td>NSLS X4A</td>
</tr>
<tr>
<td>wavelength (Å)</td>
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</tr>
<tr>
<td>space group</td>
<td>P2(_1)</td>
</tr>
<tr>
<td>no. of molecules per asymmetric unit</td>
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<tr>
<td>unit cell parameters</td>
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</tr>
<tr>
<td>(a) (Å)</td>
<td>152.72</td>
</tr>
<tr>
<td>(b) (Å)</td>
<td>62.93</td>
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<td>(c) (Å)</td>
<td>152.74</td>
</tr>
<tr>
<td>(β) (deg)</td>
<td>90.01</td>
</tr>
<tr>
<td>resolution (Å)</td>
<td>40–1.6 (1.66–1.60)</td>
</tr>
<tr>
<td>no. of unique reflections**</td>
<td>360927 (24807)</td>
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<tr>
<td>completeness (%)**</td>
<td>94.1 (65.0)</td>
</tr>
<tr>
<td>(R_{free})**</td>
<td>0.087 (0.419)</td>
</tr>
<tr>
<td>average (I/σ)</td>
<td>18.7 (3.9)</td>
</tr>
</tbody>
</table>

**Structure Determination and Model Refinement.** The structure of the Lmo2620/Zn\(^{2+}\)/PO\(_4\)^{3−} crystalline complex was determined by molecular replacement with BALBES,\(^{44}\) using the phosphotriesterase homology protein from *E. coli* (PDBe entry 1BF6) as the search model. The initial model was subjected to several iterative cycles of refinement and model building with COOT,\(^{45}\) PHENIX,\(^{46}\) and ARP.\(^{47}\) The model of the Lmo2620 Zn\(^{2+}\)-PO\(_4\)^{3−} complex was refined at 1.6 Å with an \(R_{cryst}\) of 0.178 and an \(R_{free}\) of 0.197. The final model contains six protein molecules organized in two trimers. Each protein molecule contains two well-defined zinc ions and two bound phosphates. The final crystallographic refinement statistics for the Lmo2620 Zn\(^{2+}\)-PO\(_4\)^{3−} complex are provided in Table 1.

**Computational Docking to Lmo2620.** Concurrent with the in vitro experiments, a computational strategy was used to predict substrates for Lmo2620. The high-energy intermediate (HEI) version of the Kyoto Encyclopedia of Genes and Genomes (KEGG) database of biogenic compounds was docked to the active site.\(^{48}\)\(^{49}\) A HEI is a transiently stable structure that resembles the transition state of a substrate in both structure and charge distribution. HEI states were calculated as described previously,\(^{14}\) resulting in approximately 22500 different high-energy forms for the 4207 molecules in KEGG that are candidates for one of the reactions known to be catalyzed by a member of the amidohydrolase superfamily. Because KEGG contains few phosphorylated lactones, a dedicated library was generated in a second step. This library consisted of all phosphorylated 1,4- and 1,5-lactones originating from pentose and hexose sugars, and the HEI forms were generated as before.

Docking was conducted with DOCK3.5.5\(^4\) as described previously.\(^5\) The computed poses were subjected to a distance cutoff to make sure that the O\(^−\) of the HEI portion of the molecule is found within 4 Å of the metal ions in the binding site. Of the molecules that satisfied this constraint, the top 500 compounds ranked by DOCK score (consisting of van der Waals, Poisson–Boltzmann electrostatic, and ligand desolvation penalty terms) were inspected visually to ensure the compatibility of the pose with the amidohydrolase reaction mechanism.

### RESULTS

**Cloning, Expression, and Purification of Lmo2620 and Bh0225.** The genes for Lmo2620 and Bh0225 were successfully cloned. Lmo2620 expressed well in *E. coli* in the presence of added divalent metal ions, and the protein was purified to homogeneity. The metal content of the purified proteins was measured by ICP-MS, and the results are listed in Table 2. Each enzyme contained 2.0 ± 0.1 equiv of metal, and

**Table 2. Metal Content of Lmo2620 and Bh0225**

<table>
<thead>
<tr>
<th>enzyme</th>
<th>Zn</th>
<th>Mn</th>
<th>Co</th>
<th>Fe</th>
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<td>[Zn/Zn]·Lmo2620</td>
<td>1.9</td>
<td>−</td>
<td>−</td>
<td>0.2</td>
</tr>
<tr>
<td>[Mn/Mn]·Lmo2620</td>
<td>0.2</td>
<td>1.8</td>
<td>−</td>
<td>0.1</td>
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<tr>
<td>[Co/Co]·Lmo2620</td>
<td>0.2</td>
<td>−</td>
<td>1.7</td>
<td>0.2</td>
</tr>
<tr>
<td>[Zn/Zn]·Bh0225</td>
<td>1.8</td>
<td>−</td>
<td>−</td>
<td>0.1</td>
</tr>
<tr>
<td>[Mn/Mn]·Bh0225</td>
<td>0.2</td>
<td>1.6</td>
<td>−</td>
<td>0.2</td>
</tr>
<tr>
<td>[Co/Co]·Bh0225</td>
<td>0.1</td>
<td>−</td>
<td>2.0</td>
<td>−</td>
</tr>
</tbody>
</table>

in each case, the metal added to the growth medium dominated. Minor amounts of iron and/or zinc were also found. The identity of Lmo2620 was confirmed by sequencing the first five amino acids from the N-terminus of purified [Zn/Zn]-Lmo2620. The amino acid sequence (SFIRT) matched that of MSFIRT predicted from the DNA sequence. Bh0225 was expressed in good yield in *E. coli* in the presence of added divalent metal ions. The metal content of the protein samples were measured by ICP-MS, and the results are presented in Table 2.

**Three-Dimensional Structure of Lmo2620.** The three-dimensional structure of [Zn/Zn]-Lmo2620 was determined to a resolution of 1.6 Å. Lmo2620 adopts a distorted (β/α)-β-barrel fold with N- and C-terminal extensions (Figure 2). The following chain segments are included in the eight β-strands of the barrel: β-1 (residues 18−28), β-2 (residues 61−64), β-3 (residues 87−94), β-4 (residues 149−156), β-5 (residues 183−187), β-6 (residues 210−214), β-7 (residues 234−237), and β-8 (residues 267−270). The N-terminal extension includes a β-loop (residues 3−12), and the C-terminal extension includes two distorted α-helices (residues 290−305 and 311−325). The last helix closes the entrance to the barrel from the N-terminal side. The segment connecting strands β-1 and β-2 of the barrel contains two helices, and the long chain segment between strands β-3 and β-4 contains four helices. The remaining segments between the β-strands of the barrel each contain a single helix.

The active site of Lmo2620 is located at the C-terminal end of the barrel and is open to bulk solvent. Two zinc ions are bound in the active site, with the most deeply buried zinc ion
and the most solvent-exposed zinc ion denoted as α and β, respectively. Znα is coordinated by His-22 and His-24 from strand β-1, Asp-272 from strand β-8, and the carboxylated Lys-154 from strand β-4 as illustrated in Figure 3. Znβ is coordinated by His-187 from strand β-5, His-215 from strand β-6, and the carboxylated Lys-154. The two zinc ions are bridged by the carboxylated Lys-154 and a phosphate (proximal) that acts as an additional bridging ligand. The two are bridged by the carboxylated Lys-154 and a phosphate strand β-1, Asp-272 from strand β-8, and the carboxylated Lys-154 from strand β-4 as illustrated in Figure 3. Znβ is coordinated by His-22 and His-24 from strand β-1, Asp-272 from strand β-8, and the carboxylated Lys-154 from strand β-4 as illustrated in Figure 3. Znβ is also coordinated by O2 of the bridging phosphate at a distance of 2.3–2.5 Å. Another phosphate (distal) is present at a position near the active site with a P–P distance to the proximal phosphate of 7.1 Å. There is a water molecule situated between the two phosphates at a distance of 2.5 Å to an oxygen atom of the proximal phosphate and 3.9 Å to an oxygen of the distal phosphate. The shortest distance between the oxygen atoms of the two phosphates is 4.9 Å. The distal phosphate participates in hydrogen bonds with Lys-242, Lys-244, Arg-275, and Tyr-278, as well as an interaction with Arg-218 that is mediated by two water molecules. The distance between the guanidinium amino group of Arg-218 and the distal phosphate is 4.8 Å. The presence of the distal phosphate suggests that Lys-242, Lys-244, Arg-275, Tyr-278, and Arg-218 form a very basic cavity that recognizes substrates possessing a phosphate or carboxylate group.

Docking the HEI Metabolite Library. The KEGG library contains 4207 metabolites with functional groups recognized by enzymes of the amidohydrolase superfamily. These compounds were docked against the three-dimensional structure of Lmo2620 in an average of 177707 conformations and 53990 configurations, with a total calculation time of 362 central processing unit days. Of the top 500 ranking poses, less then 300 were compatible with the recognized reaction mechanism, and five chemotypes were prominent. In order of docking rank these were lactam hydrolysis, acyclic deamination, acyclic ester and amide hydrolysis, lactone hydrolysis, and exocyclic deamination (Table 3). Detailed examination revealed incomplete fits for most of these compounds. For instance, most of the putative substrates undergoing deamination and amide and ester hydrolysis extended past the active site of Lmo2620, stranding a substantial part of the molecule in the bulk solvent. More attractive were the lactones among the top hits, most of which (34 of 46 hits) had every one of their polar atoms complemented by a corresponding interaction of an amino acid with the enzyme. Considering only those molecules that were catalytically competent and that did not strand functionality, these lactones constituted 34 of 59 of the well-placed metabolites (57.6%, a 10-fold enrichment over the 5.4% of lactones in KEGG) (Table 3 and Figure 4). Even without being filtered, they made up 46 of the top 500 dock-ranked metabolites. For comparison, there are only four lactam HEIs among the well-positioned poses (of 815 in the original library). The favorable fits and enrichment of the lactones are also borne out by a ranking that is weighted by the heavy atom count (HAC) of the small molecules; the lactones were typically smaller than many of the other molecules recognized, but more ligand-efficient, and so were among the very top-ranked molecules when the docking score was normalized for molecular size (Figure S1 of the Supporting Information). Intriguingly, 14 of the lactones were anionic, typically positioned with a carboxylate interacting with Arg-275 and other residues in the structure of Lmo2620 arranged to interact with the distal phosphate (Figure 4B). This observation turned out to be consistent with in vitro experiments that were already underway, strengthening our confidence in the importance of the lactone chemotype and the importance of a negative charge in this region of the active site.

Substrate Specificity and Kinetic Measurements. Lmo2620 was initially tested for its ability to catalyze the hydrolysis of organophosphate esters. No catalytic activity was detected with paraoxon (diethyl 4-nitrophenyl phosphate), ethyl 4-nitrophenyl phosphate, bis(4-nitrophenyl) phosphate, or 4-nitrophenyl phosphate. These results demonstrate that this enzyme is not a generic phosphotriesterase as currently annotated by NCBI. Additionally, Lmo2620 was screened with an array of lactones (Table S1 of the Supporting Information) that have previously been shown to be substrates for enzymes from groups 3.30–31, 7.15, and 9.7 of cog1735 (Figure 1), but no activity was detected with any of these compounds. The most interesting hint for potential substrates of Lmo2620 was the binding of two phosphates in the active site (Figure 3). The phosphate proximal to the metal center is most likely a surrogate for the tetrahedral intermediate formed during substrate hydrolysis, whereas the phosphate distal to the metal center likely mimics a portion of a phosphorylated (or carboxylated) substrate. Combining this clue with the preponderance of lactone substrates for related proteins from cog1735 suggested the likelihood of phosphorylated lactones as potential substrates. The
high-ranking docking compounds did not include lactone phosphates as these are not in the KEGG library. However, the appearance of anionic sugar lactones in the docking results strengthened this conviction.

Initially, eight lactones were synthesized from the monoacid derivatives of the D- and L-pentoses and then phosphorylated (Scheme 1). Of these eight compounds, only D-lyxonolactone-5-phosphate ([1]) and L-ribonolactone-5-phosphate ([8]) proved to be substrates. The stereochemistries of the hydroxyl substituents at C2 and C3 of the two substrates are identical. However, the stereochemistry of the phosphorylated substituent at C4 is either R or S for catalytic activity. Lmo2620 was also assayed with 2-deoxy-D-ribonolactone-5-phosphate ([9]) and L-ascorbate-6-phosphate ([10]), but no catalytic activity was found for either of these two compounds. The unphosphorylated sugar lactones and the diacid sugar lactones (compounds [2]–[22] in Scheme 2) were not substrates.

Bh0225 was also screened with the same set of potential substrates. No catalytic activity was detected with any of the organophosphates, unphosphorylated lactones, or diacid sugar lactones. Bh0225 preferentially hydrolyzes D-lyxonolactone-5-phosphate ([1]) and L-ribonolactone-5-phosphate ([8]) but none of the other phosphorylated lactones (compounds [2]–[7], [9], and [10]). In addition to these compounds, we also synthesized D-mannono-1,5-lactone-6-phosphate, but no hydrolytic activity could be detected by 31P NMR upon addition of Bh0225. This compound was synthesized enzymatically using yeast hexokinase and MgATP to phosphorylate D-mannono-1,5-lactone. The kinetic parameters for the metal-substituted forms of Lmo2620 and Bh0225 were determined, and the results are presented in Table 4. [Co/Co]-Lmo2620 is the most active enzyme with \( k_{\text{cat}}/K_m \) values of \( \sim 10^5 \) M\(^{-1}\) s\(^{-1}\) for the hydrolysis of compounds [1] and [8]. [Co/Co]-Bh0225 is the most active form of this enzyme, and the \( k_{\text{cat}}/K_m \) values for the hydrolysis of compounds [1] and [8] exceed \( 10^5 \) M\(^{-1}\) s\(^{-1}\).

31P NMR Analysis. NMR spectroscopy was used to examine the enzyme-catalyzed hydrolysis of compound [1] to confirm that the lactone ring was hydrolyzed rather than the monophosphate ester. The hydrolysis of compound [1] using [Co/Co]-Bh0225 was monitored as a function of time at pH 7.1 by 31P NMR spectroscopy. The substrate has a single 31P resonance at 4.03 ppm, and the D-lyxonate-5-phosphate product has a single resonance at 3.72 ppm. The changes in the 31P NMR spectra are shown in Figure 5. The resonance for

Table 3. Results of Docking the HEI Version of KEGG to Lmo2620

<table>
<thead>
<tr>
<th>Reaction type</th>
<th>Representative structures</th>
<th>N_{comp}(^{a})</th>
<th>ranks(^{b})</th>
<th>N_{int}(^{c})</th>
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</thead>
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<tr>
<td>lactam opening</td>
<td></td>
<td>54</td>
<td>2-497</td>
<td>4</td>
</tr>
<tr>
<td>acyclic deamination</td>
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<td>26</td>
<td>3-487</td>
<td>1</td>
</tr>
<tr>
<td>ester cleavage</td>
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<td>37</td>
<td>9-488</td>
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<td>16-500</td>
<td>13</td>
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<tr>
<td>lactone opening</td>
<td></td>
<td>46</td>
<td>34-491</td>
<td>34</td>
</tr>
<tr>
<td>exocyclic deamination</td>
<td></td>
<td>5</td>
<td>278-357</td>
<td>0</td>
</tr>
</tbody>
</table>

\(^{a}\)Number of molecules of a particular chemotype with a catalytically competent docking pose among the top 500. \(^{b}\)Highest and lowest rank among the top 500. \(^{c}\)Number of molecules in catalytically competent poses that have all of their polar interactions satisfied.
the substrate at 4.03 ppm decreased, and the resonance for the product at 3.72 ppm increased until all of the substrate was consumed. No inorganic phosphate or any other species was observed. These results demonstrated that Bh0225 is a lactonase and hydrolyzed compound 1 to D-lyxonate-5-phosphate.

Computational Docking to Active Site of Lmo2620.

We explored the detailed binding determinants of the two substrates, compounds 1 and 8, by docking their HEI structures. In the docked pose of compound 1, the phosphate group interacts with Arg-275, while both hydroxyl groups hydrogen bond with Lys-97. For compound 8, on the other hand, the C2 hydroxyl interacts with Lys-97 while the C3 hydroxyl interacts with Tyr-71. As with compound 1, the phosphate of 8 interacts with Arg-275 (Figure 6A,B). The poses of both compounds are positioned such that the reactive center of the HEI corresponds to an attack of the catalytic hydroxide from the si face of the lactone ring.

### DISCUSSION

Catalytic Functions in Cog1735. Enzymes within cog1735 previously have been shown to catalyze the hydrolysis of lactones and organophosphate triesters. Here, the substrate profiles for two previously uncharacterized proteins

<table>
<thead>
<tr>
<th>substrate</th>
<th>enzyme</th>
<th>$k_{cat}$ (s$^{-1}$)</th>
<th>$K_m$ (mM)</th>
<th>$k_{cat}/K_m$ (M$^{-1}$ s$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>[Zn/Zn]-Lmo2620</td>
<td>1.1 ± 0.1</td>
<td>0.46 ± 0.05</td>
<td>(2.4 ± 0.3) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Mn/Mn]-Lmo2620</td>
<td>1.7 ± 0.1</td>
<td>0.23 ± 0.02</td>
<td>(7.4 ± 0.7) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Co/Co]-Lmo2620</td>
<td>15 ± 1</td>
<td>0.25 ± 0.03</td>
<td>(6.1 ± 0.7) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Zn/Zn]-Bh0225</td>
<td>1.8 ± 0.05</td>
<td>0.40 ± 0.04</td>
<td>(4.5 ± 0.5) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Mn/Mn]-Bh0225</td>
<td>1.6 ± 0.04</td>
<td>0.20 ± 0.02</td>
<td>(8.0 ± 0.8) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Co/Co]-Bh0225</td>
<td>24 ± 1</td>
<td>0.24 ± 0.02</td>
<td>(1.0 ± 0.1) × 10$^5$</td>
</tr>
<tr>
<td>8</td>
<td>[Zn/Zn]-Lmo2620</td>
<td>1.5 ± 0.1</td>
<td>0.43 ± 0.03</td>
<td>(3.5 ± 0.3) × 10$^5$</td>
</tr>
<tr>
<td></td>
<td>[Mn/Mn]-Lmo2620</td>
<td>3.6 ± 0.1</td>
<td>0.45 ± 0.04</td>
<td>(8.0 ± 0.7) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Co/Co]-Lmo2620</td>
<td>18 ± 1</td>
<td>0.24 ± 0.03</td>
<td>(7.3 ± 0.6) × 10$^4$</td>
</tr>
<tr>
<td></td>
<td>[Zn/Zn]-Bh0225</td>
<td>1.6 ± 0.1</td>
<td>0.32 ± 0.02</td>
<td>(5.0 ± 0.5) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Mn/Mn]-Bh0225</td>
<td>1.8 ± 0.1</td>
<td>0.21 ± 0.02</td>
<td>(8.6 ± 0.9) × 10$^3$</td>
</tr>
<tr>
<td></td>
<td>[Co/Co]-Bh0225</td>
<td>30 ± 1</td>
<td>0.2 ± 0.02</td>
<td>(1.5 ± 0.15) × 10$^5$</td>
</tr>
</tbody>
</table>
from group 5 of cog1735 (Lmo2620 and Bh0225) were determined through a collective strategy of bioinformatics, docking, three-dimensional structure determination, and library screening. At the initiation of this investigation, Lmo2620 was annotated as a phosphotriesterase family protein and Bh0225 was classified as a hypothetical protein by NCBI. No further functional information was available for any of the remaining proteins from group 5 of cog1735 (Figure 1).

**Interrogation of Lmo2620 and Bh0225.** The three-dimensional structure of Lmo2620 was determined to a resolution of 1.6 Å as a necessary first step in our structure-based approach to deciphering enzyme specificity. The overall fold of this protein is that of a distorted TIM barrel, and the active site is articulated by a binuclear metal center with a bridging carboxylated lysine that is a hallmark of the amidohydrolase superfamily.22 In Lmo2620, there are two additional phosphates bound at or near the active site. The proximal phosphate bridges the two divalent cations and displaces the hydroxide that usually coordinates the two metal ions.22 The distal phosphate is ~7.1 Å from the proximal phosphate and forms polar interactions with Lys-242, Lys-244, Arg-275, and Tyr-278. These four residues are invariant within group 5 of cog1735. This observation, and the preponderance of lactone substrates for other proteins in cog1735, formed the primary structural basis for our initial prediction that Lmo2620 and Bh0225 would catalyze the hydrolysis of phosphorylated sugar lactones. Concurrently, the HEI form of KEGG had been docked against the X-ray structure of Lmo2620. The high level of enrichment of anionic lactones when the HEI form of KEGG was docked against the X-ray structure of Lmo2620 further strengthened our conviction. However, phosphorylated lactones were not observed in the initial docking calculations, because of their absence from the library.

**Substrate Identification.** Two excellent substrates were discovered for Lmo2620 and Bh0225. These compounds, D-lyxonolactone-5-phosphate (1) and L-ribonolactone-5-phosphate (8), have $k_{cat}/K_{m}$ values of approximately $10^5$ M$^{-1}$ s$^{-1}$. There is a requirement for a hydroxyl group at C2 with an $S$ configuration and a hydroxyl group at C3 with an $R$ configuration. The substituent at C4 can be in either the $R$ or the $S$ configuration. Unfortunately, we were unable to synthesize comparable phosphorylated $\gamma$- or $\delta$-lactones made from hexonic acids, and thus, it cannot be determined at this time whether these compounds would also function as substrates for these enzymes. However, we determined that the carboxylate mimics shown in Scheme 2 are not substrates for these two enzymes.

**Model for Substrate Binding in the Active Site.** All of the structurally characterized members of the amidohydrolase superfamily catalyze the cleavage of esters and amides from the same relative face of the functional group that is hydrolyzed.22,42 For esters of the type hydrolyzed by Lmo2620, it is the $si$ face of the lactone that is attacked by the bridging hydroxide.17 This stereochemical preference is recapitulated by the docking: the computed poses of compounds 1 and 8 feature an intermediate that corresponds to an attack on the $si$ face of the substrate. In this pose, the phosphate moiety at C5 is positioned such that it interacts with the same residues as the distal phosphate visible in the X-ray structure of Lmo2620. The hydroxy groups interact with Lys-97 and Tyr-71 for both substrates (Figure 6). At the same time, there is sufficient flexibility in the active site to allow the phosphate moiety of substrates with either stereochemistry at C4 to interact with the phosphate binding residues.

From a methodological standpoint, certain caveats bear mentioning. Perhaps most importantly, whereas the strategy of docking HEI forms of the KEGG metabolite library did identify what ultimately turned out to be the correct reactive center, the hydrolysis of sugar lactones, and even identified anionic forms of the lactones, it failed to identify what were ultimately the correct substrates. Whereas, in retrospect, sugar lactone phosphates do rank highly and would have been highly enriched in the docking screen, they were not in the original library.

![Figure 6. Docked poses of the substrates with Lmo2620. The position of the phosphate ion from the original structure is colored cyan; the protein and ligands are otherwise colored as in Figure 4. (A) Compound 1. (B) Compound 8.](image-url)
This highlights both the relative novelty of the activity discovered and an ongoing challenge with docking a preexisting metabolite library such as KEGG. As long as we depend upon substrates of enzymes of unknown activity being known, at full atomic detail, in KEGG, we will miss specific chemotypes. We cannot currently find substrates that are not already within KEGG by this approach. For many enzymes of unknown function, this will likely be a serious liability, meriting further effort to overcome it.

Comparison of the Structures of Lmo2620 and Dr0930. Dr0930 (PDB entries 3FDK and 3HTW) was previously identified from cog1735 as an enzyme that efficiently hydrolyzes simple δ- and γ-lactones with an alkyl substitution on the carbon adjacent to the ring oxygen. Lmo2620 and Dr0930 are ~30% identical in sequence, and their structures are highly similar. The major difference between these two proteins is the sequence (Figure 7) and conformation of the loops that follow the eight β-strands. The four residues that are important for the recognition of the phosphate moiety in Lmo2620 (Lys-242, Lys-244, Arg-275, and Tyr-278) are replaced with four hydrophobic residues (Leu-231, Met-234, Trp-267, and Leu-270) in Dr0930. These features are consistent with the observations that Lmo2620 hydrolyzes lactones with a phosphate moiety while Dr0930 hydrolyzes lactones with a long hydrophobic tail. Conservation of these residues will be a significant clue in the ongoing assault on the substrate specificity for the remaining enzymes of unknown function in cog1735.

Possible Metabolic Role of Lmo2620 and Bh0225. The specific metabolic role of Lmo2620 and Bh0225 in their respective organisms has not been established. The two substrates for these enzymes are D-lyxonolactone-5-phosphate and L-ribonolactone-5-phosphate. The sugar precursors to these compounds, D-lyxose and L-ribose, respectively, are relatively rare sugars and not abundant in nature. However, it has been reported that a mutant strain of E. coli K12 can grow on D-lyxose as the sole carbon and energy source. It is also known that Cohnella laevoribosii RI-39 can grow in a defined medium with L-ribose as the sole carbon source. We speculate that Lmo2620 and Bh0225 may be involved in a pentose (or hexose) phosphate pathway that is initiated by the phosphorylation of the lactone followed by the hydrolysis of the lactone to the acid sugar phosphate. Acid sugars of this type can be

Figure 7. Amino acid sequence alignment of selected proteins from cog1735. Lmo2620 (group 5), Bh0225 (group 5), Dr0930 (group 7), phosphotriesterase homology protein (group 1) from E. coli (locus tag b3379), and MS53_0025 from Mycoplasma synoviae S3 (group 6). The metal binding ligands are highlighted in red. The residues that have been shown to facilitate the binding of substrates in the active site of Lmo2620 and Bh0225 are highlighted in green. The corresponding residues in the other three proteins are highlighted in gray.
dehydrated to a keto sugar and subsequently cleaved via an aldolase or ketolase.

**Strategy for Functional Annotation.** Bioinformatics, three-dimensional structure determination, computational docking, and library screening were used synergistically to successfully identify two novel substrates for enzymes of unknown function in cog1735. The identification of two phosphates in the active site of Lmo2620 was leveraged with the knowledge that most of the functionally annotated enzymes within cog1735 hydrolyze lactone substrates. The preponderance of lactones in the docking calculations led to the synthesis of phosphorylated lactones and the identification of two excellent substrates for Lmo2620 and Bh0225. On the basis of the conserved residues in the active site of Lmo2620 that appear critical for substrate recognition (Tyr-71, Lys-97, Lys-242, Lys-244, Arg-275, and Tyr-278), we now predict that additional members of group 5 of cog1735 will have the same substrate profile. These proteins are listed in Table S2 of the Supporting Information.

It is also possible to predict some of the substrate attributes for enzymes of unknown function in groups 1 and 6 of cog1735 (Figure 1). MSS3_0025 from *Mycoplasma synoviae* 53 (gill 71894050), a representative from group 6, has the five conserved residues (except for two lysine/arginine substitutions) from group 5 that appear to form the phosphate binding site. This strongly suggests that this enzyme will also utilize a phosphorylated substrate. For the phosphotriesterase homology protein (PHP) from group 1, the situation is less clear because there are multiple substitutions in the phosphate binding site (Figure 7). Nevertheless, this information provides significant leverage for the successful annotation of the remaining proteins in cog1735.

### ASSOCIATED CONTENT

#### Supporting Information

A figure showing the enrichment of the most frequently observed reaction types from the docking of the HEI KEGG database to the structure of Lmo2620 (Figure S1), a list of additional lactones tried as potential substrates (Table S1), and a list of proteins expected to have the same substrate profile as Lmo2620 and Bh0225 (Table S2). This material is available free of charge via the Internet at http://pubs.acs.org.

#### Accession Codes

The X-ray coordinates and structure factors for Lmo2620 have been deposited in the Protein Data Bank as entry 3PNZ.

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**Notes**

The authors declare no competing financial interest.

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### REFERENCES


