

The Hunt for 8-Oxoguanine Deaminase

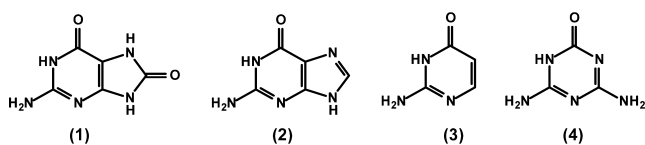
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In nature, everything that is made is unmade. Thus, for every pathway for the biosynthesis of a given metabolite, there is a complementary transformation for degradation and recycling of that building block back into living organisms. This phenomenon is particularly important for those compounds that interfere with reproduction and/or the maintenance of genetic information. One such compound is 8-oxoguanine (8-oxoG), which is formed by the oxidation of guanine within DNA by reactive oxygen species (Scheme 1).¹ If uncorrected, this modification leads to the incorporation of 8-oxoG:A mismatches and eventual G:C to T:A transversions.² Such mutagenic lesions can be excised from DNA by 8-oxoguanine-DNA glycosylase to liberate 8-oxoG as the free base.³ How then is 8-oxoG further metabolized? In many organisms, guanine is first deaminated to xanthine and then oxidized to uric acid. However, 8-oxoG is not a substrate for any of the known guanine deaminases, so this compound cannot enter the established routes for purine degradation. Chemically, the most likely transformation is the deamination of 8-oxoG directly to uric acid, but to the best of our knowledge, an enzyme catalyzing this reaction has not been discovered to date. This communication describes the systematic identification and structural elucidation of a bacterial enzyme that efficiently deaminates 8-oxoG directly to uric acid.

Scheme 1



The amidohydrolase superfamily (AHS) contains most of the enzymes known to catalyze the deamination of aromatic bases.⁴ Well-known examples include adenosine, cytosine, and guanine deaminases. The hallmark for these enzymes within the AHS is an HxxE motif that is found at the C-terminus of β -strand 5 within the context of a $(\beta/\alpha)_8$ structural fold. Approximately 1300 unique protein sequences bearing this signature can be identified in the cluster of orthologous groups as cog0402 from the completely sequenced bacterial genomes. In a search for an undiscovered subcluster of enzymes that has the ability to deaminate 8-oxoG, it was logical to assume that the recognition elements for the binding of this substrate are similar to those for the association of guanine to guanine deaminase. The fully conserved residues that directly contact guanine/xanthine in the active site of human guanine

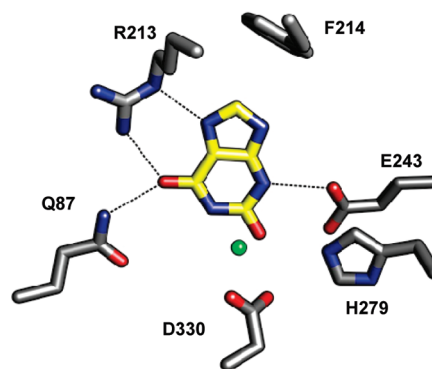


Figure 1. Active-site structure of human guanine deaminase with the product xanthine (PDB entry 2uz9). Zinc is denoted as a green sphere, and xanthine is shown with yellow carbon atoms.

deaminase are shown in Figure 1.⁵ In this complex, Asp-330, Glu-243, and His-279 form a catalytic triad of residues that are responsible for proton transfers from the activated water molecule to the two reaction products, xanthine and ammonia. In addition, the O6 carbonyl group of xanthine hydrogen bonds with the side chain of Gln-87, and the guanidino group of Arg-213 hydrogen bonds with O6 and N7 of the substrate. The phenyl ring of Phe-214 caps the size of the binding site.

The largest subcluster of AHS enzymes within cog0402 that has a global BLAST e-value of less than 10^{-70} for which the catalytic function is unknown contains ~ 200 unique protein sequences. Members of this subcluster share the conserved catalytic triad of residues found in the active site of guanine deaminase and also have the conserved glutamine residue at the C-terminus of β -strand 1. However, the conserved arginine and phenylalanine residues at the C-terminus of β -strand 4 in guanine deaminase are missing and are replaced with cysteine and serine, respectively. This subcluster of sequences in cog0402 became the prime target for the discovery of an enzyme capable of deaminating 8-oxoG. To experimentally address this proposal, one of the proteins within this subcluster of sequences (Pa0142 from *Pseudomonas aeruginosa* PA01, gi9945972) was expressed and purified to homogeneity, and its reaction profile was interrogated with a library of potential substrates.

Pa0142 was assayed against a library of more than 50 compounds with the potential for deamination of an aromatic base. Compounds in this library included pyrimidines, purines, and pterins as simple bases plus nucleosides and nucleotides. Formation of ammonia was monitored by a coupled enzyme assay using glutamate dehydrogenase, NADH, and α -ketoglutarate at pH 8.0.⁶ Product formation in each case was confirmed by mass spectrometry and UV spectroscopy. Of the compounds tested, activity could only be

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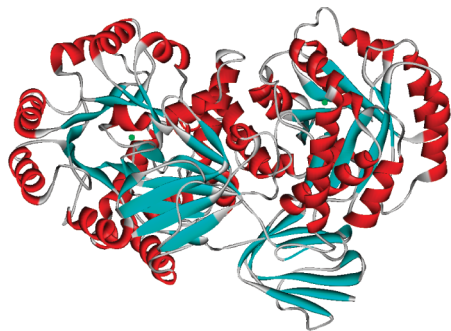


Figure 2. Ribbon representation of 8-oxoguanine deaminase (PDB entry 3h4u).

detected with 8-oxoG (**1**), guanine (**2**), isocytosine (**3**), and ammeline (**4**) (Scheme 1). The limited solubility of 8-oxoG prevented an accurate measurement of k_{cat} and K_m , but the k_{cat}/K_m value of $(2.0 \pm 0.1) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$ is an order of magnitude larger than that for the next best compound tested. For **2**, the values of k_{cat} , K_m , and k_{cat}/K_m are $3.3 \pm 0.6 \text{ s}^{-1}$, $2.6 \pm 0.6 \text{ mM}$, and $(1.3 \pm 0.4) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$, respectively. For **3**, the corresponding values are $3.9 \pm 0.1 \text{ s}^{-1}$, $1.8 \pm 0.1 \text{ mM}$, and $(2.2 \pm 0.1) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$. These results clearly support the conclusion that Pa0142 is an 8-oxoguanine deaminase (8-OGD).

The ability of Pa0142 to deaminate guanine prompted a reassessment of whether prototypical guanine deaminases can utilize 8-oxoG as a substrate. Two putative guanine deaminases were cloned, purified, and characterized: Pa0134 from *P. aeruginosa* (gil15595332) and b2883 from *Escherichia coli* K12 (gil16130785). These enzymes efficiently catalyze the deamination of guanine but not any other compounds. For Pa0134, the values of k_{cat} , K_m , and k_{cat}/K_m for guanine are $21 \pm 2 \text{ s}^{-1}$, $0.50 \pm 0.09 \text{ mM}$, and $(4.2 \pm 0.9) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$, respectively. For b2883, the corresponding values are $12.5 \pm 0.2 \text{ s}^{-1}$, $0.15 \pm 0.01 \text{ mM}$, and $(8.1 \pm 0.4) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$. The maximal rate constant for the deamination of 8-oxoG is less than $5 \times 10^{-3} \text{ s}^{-1}$.

Attempts to crystallize Pa0142 have thus far failed. However, a homologous protein identified from environmental DNA sequencing efforts in the Sargasso Sea (gil44264246) was purified and crystallized and its three-dimensional structure determined by X-ray diffraction methods. This protein is 43% identical in protein sequence to Pa0142, and it is 98% identical to a protein of unknown function from *Burkholderia* sp. 383 (Bcep18194_A5267). The gene was codon-optimized for expression in *E. coli*, synthesized, and TOPO (Invitrogen)-cloned into a vector that yields a protein with a noncleavable histidine tag. The sequence and expression/purification details are available from the PepcDB database (<http://pepcdb.pdb.org/>) under targetID "NYSGXRC-9236e", and the plasmid is available from the PSI Material Repository (psimr.asu.edu). The protein (6.2 mg/mL) crystallized in the presence of 0.5 mM ZnCl_2 and 2.8 M sodium acetate, and the structure was determined to a resolution limit of 2.2 Å using data collected at the NSLS X4A beamline at Brookhaven National Laboratory (PDB entry 3h4u; see Table S1 in the Supporting Information).

The Pa0142 homologue is more active than Pa0142 for the deamination of guanine, isocytosine, and 8-oxoG. For 8-oxoG, the value of k_{cat}/K_m is $(2.7 \pm 0.1) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$. With isocytosine as a substrate, the values of k_{cat} , K_m , and k_{cat}/K_m are $62 \pm 2 \text{ s}^{-1}$, $1.6 \pm 0.2 \text{ mM}$, and $(3.9 \pm 0.5) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$, respectively. Kinetic constants of $52 \pm 3 \text{ s}^{-1}$, $1.0 \pm 0.1 \text{ mM}$, and $(5.2 \pm 0.6) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$ were obtained for k_{cat} , K_m , and k_{cat}/K_m , respectively, using guanine as a substrate. No activity was obtained with any of the substituted pterins that were tested.

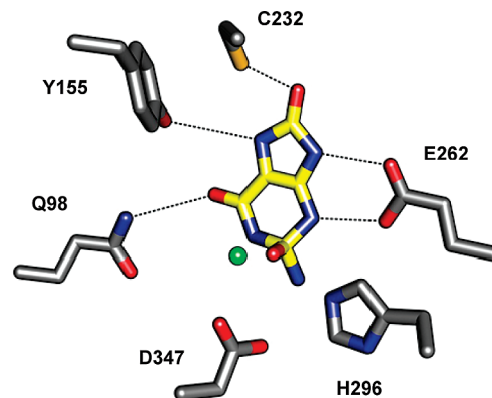


Figure 3. Active-site model for 8-oxoguanine deaminase with a putative tetrahedral intermediate for deamination of 8-oxoG.

The structure of the dimeric enzyme is presented in Figure 2. A single zinc ion is bound in the active site. The divalent cation is coordinated to two histidine residues (H93 and H95) that follow β -strand 1 and another histidine from β -strand 5 (H259). The coordination scheme is completed by the interaction of the side-chain carboxylate group of an aspartate from β -strand 8 (D347) and a water molecule. The metal ligation scheme is identical to that previously observed for cytosine, guanine, and adenosine deaminases.⁴

The Pa0142 homologue has a conserved glutamine residue (Q98) in the active site that is homologous to Q87 in human guanine deaminase. However, the fully conserved arginine and phenylalanine residues that complete the binding pocket for guanine/xanthine in the active site are missing. These residues, from the C-terminus of β -strand 4, are substituted with a cysteine (C232) and serine (S233), respectively. C232 is positioned to donate a hydrogen bond to the carbonyl group at C8 of 8-oxoG, and E262 can facilitate a proton transfer to N3 during catalysis. A fully conserved tyrosine residue (Y155) may additionally interact with N7. A computational model for these interactions with a putative tetrahedral intermediate is presented in Figure 3.^{7,8}

In summary, an enzyme from *P. aeruginosa* that efficiently deaminates 8-oxoguanine to uric acid has been discovered. A database search has identified more than 200 other organisms that possess at least one protein that is >35% identical in sequence to this enzyme (see Table S2). The three-dimensional structure of 8-OGD has been determined, and computational methods have been utilized to rationalize the binding of 8-oxoG in the active site. Experiments designed to verify the in vivo activity of Pa0142 are in progress.

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Supporting Information Available: Experimental details and Tables S1 and S2. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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